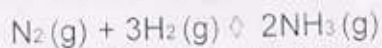


# K.R.D.A.V.PUBLIC SCHOOL

## HOLIDAY HOMEWORK CHEMISTRY XI

### Unit 1

- Q.1. Define molality. What is relation between molality and mole fraction?  
Q.2. Hydrogen reacts with  $N_2$  to produce ammonia according to the equation



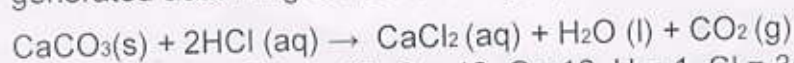
Determine how much ammonia would be produced if 200 g of  $H_2$  reacts with excess of  $N_2$ , [Atomic weight of N = 14, H = 1 u]

- Q.3. Calculate number of moles in the following cases:

(i) 7.85g of Fe (Atomic weight of Fe = 56 u)

(ii) 4.68 mg of Si (Atomic weight of Si = 28u)

- Q.4. If 2.0g of  $CaCO_3$  is treated with 2.0 g of HCl, how many grams of  $CO_2$  can be generated according to the following equation?



[Atomic weight of Ca = 40, C = 12, O = 16, H = 1, Cl = 35.5 u]

- Q.5. Calculate number of atoms in each of the following:

(i) 5.6 L of  $NH_3$

(ii) 4.4 g of  $CO_2$

(iii) 52 u of He

Atomic weight of He = 4 u

- Q.6. A compound contains 54.2% C, 9.2% H and 36.6% oxygen, determine its empirical formula if its molecular weight is 88 u. [At. Wt. Of C=12, H=1, O=16u]

- Q.7. Oxygen is prepared by catalytic decomposition of  $KClO_3$ . Calculate amount of  $KClO_3$  required to produce 24 mole of oxygen. (At. Wt. Of K = 39, Cl = 35.5, O = 16 u)

- Q.8. Concentrated  $H_2SO_4$  is 98% weight and has density 1.84 g  $cm^{-3}$ . What volume of concentrated acid is required to make 5.0 L of 0.50 M  $H_2SO_4$  solution?

- Q.9. (i) State and explain Avogadro's law. Illustrate it with an example.

(ii) 500 ml of 0.250M  $Na_2SO_4$  solution is added to an aqueous solution of 15.00 g of  $BaCl_2$  resulting in the formation of white precipitate of  $BaSO_4$ .

How many moles and how many grams of  $BaSO_4$  are formed? [Atomic weight of Ba = 137, Cl = 35.5, S = 32, O = 16u]

- Q.10. Calculate the number of molecules present in

(i) 34.2g of cane sugar

(ii) a drop of water having a mass of 0.05g.

- Q.11. Two oxides of a metal contain 27.6% and 30% of oxygen respectively. If the formula of the first oxide is  $M_2O_3$ . Find the second.

- Q.12. What is the mole fraction of methanol in solution containing 3.2 g of methanol in 18 g of  $H_2O$ .

- Q.13. Commercially available concentrated hydrochloric acid contains 38% HCl by mass.

(i) What is the molarity of this solution? The density is 1.19g/ $cm^3$ .

(ii) What volume of concentrated HCl required to make 1.00L of 1.10 M HCl? (Ans: (a) 12.38M, (b) 123.8L)

- Q.14. A solid mixture (5.0g) consisting of lead nitrate and sodium nitrate was heated below 600°C until the weight of the residue was constant. If the loss in weight is 28% find the composition of mixture. (Ans:  $Pb(NO_3)_2 = 3.32$  g.)



Q.15. Convert the following into basic units

(i) 28.7 pm (ii) 5.15  $\mu\text{m}$

Q.16. Explain Dalton's atomic theory and its drawbacks.

Q.17. On analysis a substance was found to have the following percentage K=31.84 Cl = 28.98 O = 39.18 Calculate its molecular formula if its molecular mass is 122.5

## Unit : 2

Q.1. State Pauli exclusion principle.

Q.2. Why is it not possible to determine exact position and momentum of fast moving sub-atomic particles like electrons simultaneously? Name the principle.

Q.3. Calculate the energy and frequency of light required to excite an electron in

hydrogen atom from first energy level to fourth energy level.  $h = 6.626 \times 10^{-34} \text{ Js}$ .

Q.4. The frequency of radio waves is 1368 kHz. Calculate its wavelength (in meters) and energy.

Q.5. The mass of an electron is  $9.1 \times 10^{-31} \text{ kg}$ . If its kinetic energy is  $3.0 \times 10^{-25} \text{ J}$ . calculate its wavelength.

Q.6. Explain dual nature of electrons. Give de Broglie equation.

Q.7. (i) Write electronic configuration of Cr (24) and Cu (29).

(ii) An atomic orbital has  $n=3$ . What are possible values of  $l$  and  $m_l$

Q.8. A photon of wavelength  $4 \times 10^{-7} \text{ m}$  strikes on the metal surface, the work function of metal being 2.13 eV. Calculate:

(i) Energy of photon in eV.

(ii) The kinetic energy of the emitted electron.

(iii) The velocity of photoelectron. [ $1 \text{ eV} = 1.602 \times 10^{-19} \text{ J}$ ]

Q.9. (i) Why are half filled and completely filled orbitals more stable?

(ii) Calculate the number of photons that provide 1 J of energy.

Q.10. What is atomic spectrum? Describe hydrogen spectrum and describe series of spectrum and the region to which do they belong? What is maximum number of emission lines when the excited electron of a H atom in  $n = 6$  drops to ground state.

Q.11. If  $n$  is equal to 3, what are the values of quantum numbers  $l$  and  $m$ ?

Q.12. What is the Rydberg formula? How does it help to calculate the wave numbers of different series of lines of hydrogen spectra?

Q.13. How does Bohr's model explain simultaneous appearance of a large number of lines in the hydrogen spectrum?

Q.14. How do 1s and 2s orbital differ from each other?

Q.15. List the main points of difference between orbit and orbital.

Q.16. What were the main points of electromagnetic theory? What were its limitations?

How have these been over come by Plank's quantum theory?

Q.17. Using the concept of quantum numbers, calculate the distribution of electrons as well as maximum numbers electrons in the N shell.

Q.18. How the study of quantum numbers led to Pauli's exclusion principle?

Q.19. Find the number of protons, electrons and neutrons in  $^{13}\text{Al}^{3+}$  and  $^{8}\text{O}$

Q.20. Which of the following orbital is not possible 1p, 2s, 3f, 4d?

Q.21. Write the electronic configurations?  $\text{Cr}$ ,  $\text{Cr}^{3+}$ ,  $\text{Na}^{+}$ ,  $\text{N}^{3-}$ ,  $\text{N}_2^{+}$

Q.22. What are the limitations of Bohr's model?

Q.23. What is the difference between Rydberg's formula and Balmer's formula?

Q.24. Why electronic energy is negative?

Q.25. What are quantum numbers and give the significance of each quantum number?

XI medical

Biology

Learn ch -1 to 3  
make herbarium of  
minimum 20 plants**CLASS XI-MATHS HOLIDAY'S HOMEWORK**

1. Do chapter wise assignment of ch-1.2.3.4
2. Revise same chapter from N.C.E.R.T and from EXTRA BOOK
3. Prepare your self for presentation on any one topic [following topics are]
  - a) Biography of famous mathematicians
  - b) Presentation on any topics which you have studied in class XI
4. Prepare 20 questions (10 one marks, 10 two marks) according to your group.





Kali Ram D A V Public School, Safidon

Class XI (English)

Holiday's Homework

1. READING SECTION :-

(I) Unseen passage – Evergreen Test Assignment 1 to 5 (pg. no. 16-25)

2. WRITING SECTION:-

Article Writing -Evergreen Test Assignment -19,20 (pg. no. 167,168)

Topic – “Trash Called Entertainment”

“ Delights And Usefulness of Walking”

Letter Writing-Evergreen Test Assignment- 12 (pg. no. 137,138)

Letter to the Editor

Letter to the Editor suggesting some ways and means to reduce global warming

Notice Writing :- Evergreen Test Assignment -1 (pg. n0. 73)

Topic: Special Cultural Programme on the eve of Lohri

Tour to the Rocket launching Station

3. GRAMMAR SECTION :-

DETERMINERS-Evergreen Test Assignment-27 (pg. no. 203, 204 ) (EX.1-5)

TENSES -Evergreen Test Assignment-29(pg. no.214 , 215)(EX.1-5)

VOICE -Evergreen Test Assignment-32(pg. no. 229 , 230 )(EX.1-5)

4. LITERATURE SECTION :-

Learn Lessons discussed in class

HORNBILL-Lesson- 1,2,3

Poem-1,2

SNAPSHOTS-Lesson-1,2,3

**HOLIDAYS HOMEWORK PHYSICS (XI)**

- 1) Learn all lessons.
- 2) Solve given assignments.
- 3) Practice previous year question papers.
- 4) Make the project.

Write and learn the following questions:

1. State the different type of Economic Activities.
2. What are the objectives of Business?
3. What is the Role of Profit in business?
4. Define Trade. Explain briefly different types of trade.
5. What do you understand by Sole Proprietorship? Outline its features.
6. Outline the procedure for registration of a Partnership. Write the effects of non-registration of a partnership firm.



## 7. Case <sup>Study</sup> related Questions 3

1. A famous doctor charges highly for consultation from his patients and refuses to treat the poor patients without consultation charges. He pays attention to the representatives and agents of Pharma Co. and takes gifts and commission from them also.

In your view, is it professional behaviour of a doctor? Is he following the code of conduct of doctor?

2. A group of Bourgeois friends setup a factory. This factory was set up in a residential locality. Its machines were very noisy. The children could not concentrate on their study without distraction. On the other hand, this factory provided employment to so many people.



Point out the values being affected<sup>3</sup> here. Mention any three of them.

3. Alka, Komal and Kamal are partners in a Firm. Kamal uses Rs 20 Lac of the Firm, without any information to the other partners to purchase a house and he earns a profit of ₹ Five Lac on it. Which values are being violated here? Also tell, what will happen to the profit earned by Kamal?

### Activity

Select a local enterprise and find out the objectives of that enterprise.

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**Class- X1 COMMERCE ( ACCOUNTANCY)**

**SUMMER VACTION ASSIGNMENT**

Revise the numericals and learn the very short answers of the following chapters

- (1) Accounting equations
- (2) Books of origin entry – journal
- (3) Books of origin entry – cash book

2) Learn the following chapter

- (1) Meaning and objectives of accounting
- (2) Basic accounting terms
- (3) process and bases of accounting
- (4) double entry system

3) prepare project work on source documents of accountancy contents-

- 1 cash memo
- 2 invoice and bill
- 3 receipt
- 4 debit note
- 5 credit note
- 6 pay in slip
- 7 cheque
- 8 voucher – types :

**HOLIDAY'S HOMEWORK (2019-20) CLASS:-11TH (POLITICAL SCIENCE)**

1. Prepare a Project on "LOK SABHA ELECTION IN INDIA(2019)"

2. Prepare Power Point Presentation on following :-

TOPIC	ROLL NO
i. Constitution Why and How?	1 to 7
ii. Right in the Indian Constitution	8 to 14
iii. Election and Representation	15 to 21
iv. Executive	22 to 30

3. Revise full syllabus taught in month of April and May.

4. Slow learner students read ch. 1 to 4 and learn only one and three



N K D A V Public School, Sandon  
Holiday's Homework -2019-20  
Class XI (Information Technology 802)

1. Learn Question-Answers from Unit -1 Computer Organisation & OS
2. Framing 100 Objective Type Questions from Unit -3 Office Automation Tools (Ch-3 Word Processing (Writer))
3. Design in Practical File:-
  - (a) Draw a Computer Diagram With Input Unit, Processing , Output and Storage Units.
  - (b) Design a Bio Data (Resume) in Writer Software.
  - (c) Design a Letter/Application Formatting in Writer Software.
  - (d) Design a Find and Replace Command Demo. in Writer.
  - (e) Design a Auto Text Command Demo. in Writer.

## HOLIDAY'S HOMEWORK (2019-20) CLASS:-11TH (HISTORY)

1. Prepare a Project on any one topic in the following topics:-

1. Anthropological Research based on Darwin's Theory
2. Critique of the Industrialization in Britain
3. Relations and Impacts of past crusades
4. Making and Unmaking of Mesopotamia
5. Paradigms of Greco-Roman Civilization
6. Aspirations of women in Renaissance period
7. Paths to Modernization of Japan/China
8. An Exploratory study into Humanism
9. Piecing together the past of Genghis Khan
10. An in depth study into the "now and then" paradigm of Christianity
11. An exploratory study into the realism and the transmission of Humanistic ideas
12. Scientific Revolution and the origins of modern science
13. An exploratory study into the making of America
14. Myriad Realms of Slavery in ancient, medieval and modern world
15. Learning about global Sufism
16. History of aborigines-America/Australia

2. Prepare Power Point Presentation on following :-

TOPIC	ROLL NO
1. Introduction to World History	1 to 10
2. From the Beginning of Time	11 to 20
3. Writing and City Life	21 to 30

3. Revise full syllabus taught in month of April and May.

**4. Slow learner students read ch. 1 to 4 and learn only one and three marks question –answer of covered lessons**



# ECONOMICS HOLIDAY HOMEWORK

## CLASS-11<sup>th</sup>

*1] Prepare a project file on any topic related to your syllabus or choose any topic that are given below-*

- *Food supply channel in India*
- *Goods and Services Tax Act and its impact on GDP*
- *Self – help group*
- *Digital India – Step towards the future*
- *Make in India – The way ahead*
- *Health Expenditure [of any state]*
- *Micro and small scale industries*
- *Organic Farming – Back to the nature*

*2] Learn full syllabus which have been completed in your class.*

Hindi Holiday Homework  
Class XIA<sub>4</sub>

1. Revision work:-

काव्य खंड- (पद मीरा और कबीर)

गद्यांश - पाठ नमक का दारौगा

पाठ भारतीय में गायिकाओं में बेजोड लता मंगेशकर

अभिव्यक्ति और माध्यम- पाठ जनसंचार

2. Creative work:-

Activity (1947-2019 तक सभी अखबारों की व्याख्या

अपको कौन सी फिल्म पसंद है और उसमें कौन सा किरदार पसंद है क्यों ?

(Activity work- N.B.)

3. Written work

औपचारिक पत्र (2)

अनौपचारिक पत्र (2)

निबंध लेखन:-

- 1) मोबाईल फोन वरदान या अभिशाप
- 2) सर्वशिक्षा अभियान
- 3) उपभोक्ता संस्कृति का भड़ता प्रभाव।
- 4) लोकतंत्र में मिडिया का सामाजिक दायित्व।

फीचर लेखन :-

- 1) कंप्यूटर और हम
- 2) नशे के दलदल में धस्ती नयी पीढ़ी

रिपोर्ट लेखन :-

- 1) आपके घर के बाहर खड़ी मोटरसाइकल चोरी हो गई उसकी रिपोर्ट थाने में लिखित रूप में किजिए
- 2) विद्यालय में हैड ब्याय और हैड गर्ल के चुनाव के लिए बुलाई गई सभा से संबंधित एक रिपोर्ट लिखिए।

आलेख लेखन:-

- 1) प्रात जागिए और स्वास्थ्य रहिए
- 2) हरे भरे वृक्ष।